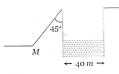
# **EW STANDARD ACADE**

## Date : 24-06-24

 $CLASS: 11^{TH}$ JEE Time: 3 HRS

## **PHYSICS**

- Why does a tennis ball bounces more at 1. hills
- 2. What is angle between velocity vector and acceleration vector in uniform circular motion?
- 3. A particle moves with a constant speed but in constantly varying direction. The path of particle will be?
- 4. The speed of a projectile at its maximum height is half of its initial speed. The angle of projection is?
- 5. Two second after projection, a projectile is travelling in a direction inclined at  $30^{\circ}$  to the horizontal, after one more sec, it is travelling horizontally the magnitude and direction of its velocity are?
- 6. A body is projected up a smooth inclined plane (length = $20\sqrt{2}$  m) with velocity u from the point M as shown in the figure. The angle of inclination is  $45^{\circ}$  and the top is connected to a well of diameter 40m. If the body just manages to cross the well, what is the value of *u*



- 7. A cricketer can throw a ball to a maximum horizontal distance of 100 m. How much high above the ground can the cricketer throw the same ball?
- 8. A particle moves along a semicircle of radius 10 m in 5 seconds. The average velocity of the particle is?
- 9. A ball is dropped downwards. After 1 second another ball is dropped downwards from the same point. What is the distance between them after 3 second?
- 10. A body freely falling from the rest has a velocity v after it falls through a height

h.The distance it has to fall down for its velocity to become double, is?

## CHEMISTRY

- 1. What are sigma ( $\sigma$ ) and pi ( $\pi$ ) bonds?
- 2. What is s-p overlapping?
- 3. Explain  $sp^2$  hybridisation with example
- 4. Explain in short, the factors affecting the formation of ionic bond.
- 5. Describe the characteristics of covalent compounds.
- 6. Write the electronic structures of PCl<sub>5</sub> and NH<sub>4</sub>Cl
- 7. What is VSEPR theory? Explain structure of  $H_2O$  with the help of it.
- 8. State the role of LCAO in the molecular orbital theory
- 9. What is hydrogen bond? What is the reason as normal temperature water is liquid and H<sub>2</sub>S is gas?
- 10. Describe the geometry of the following molecules on the basis of hybridization and VSEPR theory-
  - (i) Methane
  - (ii) Water
  - (iii) NH<sub>3</sub>.
  - (iv) PC<sub>15</sub>

# MATHS

- If  $sin(x + 3\alpha) = 3sin(\alpha x)$  then tan x is. 1.
- If tan  $\alpha$  equals the integral solution of the 2. inequality  $4x^2 - 16x + 15 < 0$  and  $\cos \beta$ equals to the slope of the bisector of first quadrant, then  $\sin(\alpha + \beta) \sin(\alpha - \beta)$  is equal to
- 3. Prove that  $\tan 70^{\circ} \tan 20^{\circ} = 2 \tan 50^{\circ}$ .
- 4. Prove that  $\frac{\sin x \sin 3x}{\sin^2 x \cos^2 x} = 2 \sin x$ . 5. Prove that  $\sin 47^0 + \sin 61^0 \sin 11^0 \sin 11^0$
- $25^0 = \cos 7^0$ .
- 6. Prove that  $\frac{\sec 8\theta 1}{\sec 4\theta 1} = \frac{\tan 8\theta}{\tan 2\theta}$
- 7. The value of the expression

 $1 - \frac{\sin^2 y}{1 + \cos y} + \frac{1 + \cos y}{\sin y} - \frac{\sin y}{1 - \cos}$  is equal to

